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A Novel Stereoselective Route to Alkyl 2-Deoxy- β -D-glucosides via S-(2-Deoxy- α -glucosyl) Phosphorodithioates

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Communication

A NOVEL STEREOSELECTIVE ROUTE TO ALKYL 2-DEOXY-
 β -D-GLUCOSIDES VIA S-(2-DEOXY- α -GLUCOSYL)
 PHOSPHORODITHIOATES

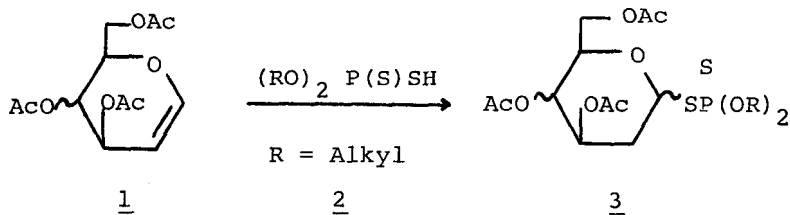
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Intense interest in sugar components of anthracycline anti-
 biotics and antitumor drugs generated a need for stereoselective
 methods of synthesis of 2-deoxy sugar glycosides. Although several
 efficient procedures leading to 2-deoxy- α -glycosides were elaborated,
 there is no satisfactory general procedure leading to the β -anomers.
 The methods described are not stereoselective and involve separation
 of anomeric intermediates of glycosides.²

In the recent studies from this laboratory it was shown that
 O,O-dialkylphosphorodithioic acids (2) add quantitatively and regio-
 specifically to the acetylated glycals (1) to give the phosphorodi-
 thioates (3).³ The stereoselectivity of the addition depends on the

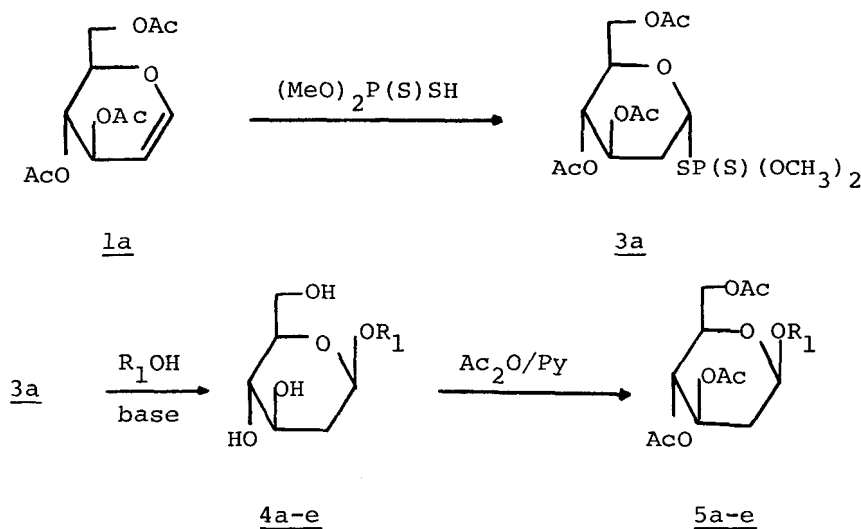


SCHEME 1

substituents at the phosphorus atom. The addition of commercially available O,O-dimethylphosphorodithioic acid proceeds in fully stereoselective manner leading exclusively to the α -anomers.

We now report a novel approach to the synthesis of alkyl 2-deoxy- β -D-glucosides starting from D-glucal via the adduct (3). Our synthetic procedure is based on the nucleophilic displacement of the dialkyldithiophosphoryl group of (3) by alcohols in the presence of bases which proceeds with full inversion of configuration at the anomeric center. Interestingly, the rate of this displacement is considerably higher than that at the phosphorus atom and according to the ^{31}P NMR evidence there is no indication of sulphur-phosphorus bond cleavage. In the case of peracetylated sugars this synthesis is accompanied by deacetylation and leads directly to the unprotected 2-deoxy- β -D-glycosides.

The synthesis of alkyl 2-deoxy- β -glycosides is exemplified by the sequence of reactions starting from 3,4,6-tri-O-acetyl-D-glucal (1a).



For R_1 , see Table 1.

SCHEME 2

The formation of the β -glucosides was quantitative according to the ^{31}P NMR data;⁴ however, in order to verify the actual yields of isolated products, the glucosides (4a - e) were reacylated. In all cases the yield of analytically pure acetylated alkyl 2-deoxy- β -D-glucosides (5a - e) exceeded 85%. The β -configuration of the glucosides obtained is evident from their ^1H NMR spectra and other properties, including the specific rotation values (Table 1).

In a typical procedure performed on a mmole scale stoichiometric amounts of freshly distilled O,O-dimethylphosphorodithioic acid and 3,4,6-tri-O-acetyl-D-glucal are dissolved in ca. 5 mL of dry benzene. The addition is completed within 48 hrs at room temperature. The reaction mixture is concentrated to a small volume, a twofold excess of the appropriate alcohol containing 1.1 equivalent of sodium is added and the mixture kept at room temperature for ca. 2 hrs. Evaporation of the solvents in vacuo affords the semi-crystalline residue which is then acetylated by acetic anhydride/pyridine according to standard procedure, and the per-acetylated alkyl 2-deoxy- β -D-arabinohexoside is isolated in the usual way.⁶

Further application of this method to the synthesis of oligo-2-deoxysaccharides, 2-deoxy-N-glycosides and 2-deoxy-2-oximino- β -D-glycosides are under way and will be reported in due course.

ACKNOWLEDGMENTS

This work was supported by the grant from the Polish Academy of Sciences (MR-I-12).

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TABLE 1 : Physical and NMR Data for alkyl 3, 4, 6-tri-O-acetyl-2-deoxy-β-D-glucopyranosides 5 (a-e) ^a

Compound	R ₁	M. p. ^b (°C)	[α] _D ^c 18°	H-1	¹ H NMR ^d δ (ppm)
<u>5a</u>	CH ₃ ^e	96 - 97	- 22	H-1	4.48 (J _{1,2a} = 9.5 Hz, J _{1,2e} = 2 Hz)
<u>5b</u>	C ₂ H ₅	80 - 81	- 28	H-1	4.58 (J _{1,2a} = 9.5 Hz, J _{1,2e} = 2 Hz)
<u>5c</u>	n-C ₃ H ₇	55 - 57	- 31	H-1	4.57 (J _{1,2a} = 9.5 Hz, J _{1,2e} = 2 Hz)
<u>5d</u>	i-C ₃ H ₇ ^f	64 - 66	- 34	H-1	4.65 (J _{1,2a} = 9.5 Hz, J _{1,2e} = 2 Hz)
<u>5e</u>	i-C ₄ H ₉	67 - 68	- 50 ^g	H-1	4.52 (J _{1,2a} = 10 Hz, J _{1,2e} = 2.5 Hz)

^a For compounds 5 (a-e) satisfactory elemental analyses were obtained. ^b From diethyl ether. ^c In chloroform. ^d ¹H NMR spectra were recorded with a Varian 60 MHz apparatus, CDCl₃ as solvent. ^e Ref. 2b and 2f. ^f Ref. 5. ^g [α]_D²⁰ ₅₇₈.

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